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HEAT-TRANSFER TESTS OF AQUEOUS ETHYLENE GLYCOL

SOLUTIONS IN AN ELECTRICALLY HEATED TUBE

By Everett Bernardo and Carroll S. Eian

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ADVANCE RESTRICTED REPORT

HEAT-TRANSFER TESTS OF AQUEOUS ETHYLENE GLYCOL

SOLUTIONS IN AN ELECTRICALLY HEATED TUBE

By Everett Bernardo and Carroll S. Eian

SUMMARY

As part of an investigation of the cooling characteristics of liquid-cooled engines, tests were conducted with an electrically heated single-tube heat exchanger to determine the heat-transfer characteristics of AN-E-2 ethylene glycol and other ethylene glycol-water mixtures for the following ranges of conditions:

Similar tests were conducted with water and commercial butanol (n-butyl alcohol) for check purposes.

The results of tests conducted at an approximately constant liquid-flow rate of 0.67 pound per second (Reynolds number, 14,500 to 112,500) indicate that at an average liquid temperature of 200° F, the heat-transfer coefficients obtained using water, nominal (by volume) 30 percent-70 percent and 70 percent-30 percent glycol-water mixtures are approximately 3.8, 2.8, and 1.4 times higher, respectively, than the heat-transfer coefficients obtained using AN-E-2 ethylene glycol.

The heat-transfer coefficients of the coolants tested were satisfactorily correlated using the following equation:

$$\frac{hD}{k} = 0.048 \left(\frac{c\mu}{k}\right)^{0.4} \left(\frac{DG}{\mu}\right)^{0.73}$$

where h is heat-transfer coefficient; D is inside diameter of tube; k is thermal conductivity of liquid; c is specific heat of liquid; μ is absolute viscosity of liquid; and G is mass rate of liquid flow. In the evaluation of this equation, the physical properties used for the aqueous ethylene glycol solutions and water were those compiled by C. S. Cragoe (National Bureau of Standards) for the Coordinating Research Council.

INTRODUCTION

A satisfactory analysis of liquid-cooled engine cooling data requires a knowledge of the heat-transfer properties of the coolants used. Heat-transfer characteristics of liquids may generally be predicted from their physical properties by means of the Nusselt relation, which has been experimentally verified for a variety of liquids (reference 1, p. 181). The physical properties of ethylene glycol and ethylene glycol-water mixtures have been experimentally determined ever limited temperature ranges and have been extrapolated beyond these ranges (reference 2). Few heat-transfer data, however, have been previously obtained for AN-E-2 othylene glycol and other ethylene glycol-water mixtures; hence, the applicability of the physical properties of these coolants for a range of temperatures to the correlation of heat-transfer coefficients by established theory has not been ascertained.

As part of an investigation of the cooling characteristics of liquid-cooled engines, the tests reported herein were conducted during the winter of 1943 and the spring of 1944 at the NACA Cleveland laboratory in order to provide the heat-transfer data required for such a correlation. Forced-convection heat-transfer coefficients were determined for AN-E-2 ethylene glycol, nominal (by volume) 70 percent-30 percent and 30 percent-70 percent glycolwater mixtures for a range of average liquid temperatures, liquid-flow rates, and heat fluxes. Heat-transfer coefficients were also determined for water and commercial butanel (n-butyl alcohel) for check purposes inasmuch as heat-transfer data for these liquids are available (reference 1, pp. 180 and 181).

The tests were conducted in a modified version of the singletube heat exchanger described in reference 3. The tube was electrically heated by the passage of current through the tube, which resulted in heat fluxes of the same order of magnitude as those prevailing in modern liquid-cooled engine cylinders.

APPARATUS

A schematic diagram of the electrically heated single-tube heat exchanger and the associated equipment used in the tests is shown in figure 1.

Heater Tube

The details of the heater-tube section, which consisted of an 18:8 stainless-steel tube with a 1/2-inch outside diameter and a 1/32-inch wall thickness, are shown in figure 2. Copper adapters were silver-soldered to each end of the tube, resulting in an effective tube length of 22.75 inches. Each adapter was connected to a 9-inch length of 1/2-inch standard pipe (0.62 in. I.D.) and a 6-inch electric-insulating coupling of the same internal diameter (fig. 1).

Tube-wall temperatures were measured at 25 locations (fig. 2) by means of iron-constantan thermocouples (24-gage flexible-glass insulated wire) and a calibrated self-balancing indicating-type potentiometer. The thermocouples were spot-welded to the outside of the tube wall and precaution was taken that the last point of contact between the wires was at the tube surface. The tube was thermally insulated by a wrapping of flexible-glass tape, a 1-inch layer of glass wool, and a 1/4-inch layer of asbestes.

Electrical System

Power was supplied to the tube from a 208-volt alternatingcurrent supply line through an autotransformer, a voltage regulator of the saturable-core reactor type, and a 20:1 power transformer (fig. 1). The electrical connections at the tube were made through clamp-type copper connectors. The tube was electrically insulated from the rest of the system by the nonconducting couplings.

The autotransformer and voltage-regulator unit permitted adjusting and maintaining a constant voltage and the power transformer provided large currents through the tube. A calibrated ammeter in conjunction with a 240:1 instrument current transformer was used to measure the current through the tube and a high-resistance calibrated voltmeter connected across the tube at the copper adapters was used to measure the voltage drop. The voltmeter leads were made of No. 8 solid copper wire and were maintained as short as possible in order to obviate voltmeter corrections.

Liquid System

The liquid was circulated by a centrifugal pump through a heating and cooling blending unit and then through a plate-type filter to the tube (fig. 1). From the tube the liquid flowed through a rotameter and back to the pump with a small amount of the liquid being shunted to a tank that was located above the highest point in the system. The tank provided for liquid expansion, makeup liquid, and the introduction of compressed air for conducting tests at liquid pressures above atmospheric. A bleed line from the tank was used to relieve the compressed air when tests were conducted at atmospheric pressure.

The liquid-flow rate through the tube was regulated by a throttle valve located at either end of the tube (fig. 1). The flow rate was measured with the rotameter, which had been calibrated for a range of temperatures with the various liquids used. The liquid temperature into the tube was controlled with the heating and cooling blending unit, which consisted of an electric heater, a cooler, and a mixing-valve-type temperature regulator. Liquid temperatures were measured at the entrance and the exit of the tube with single thermocouples in conjunction with the self-balancing indicating-type potentiometer.

More accurate measurements than those obtained with the single thermocouples were afforded by two thormopiles in combination with a portable potentiameter. The thermopile construction and the method of installation is illustrated in figure 3. Each of the two thermopiles consisted of four single thermocouples connected in series and distributed across the pipe diameter. The thermopiles were also connected differentially in order to measure directly the temperature rise of the liquid in flowing through the tube. The hot junction of all the liquid thermocouples was coated with an insulating varnish in order to reduce the possibility of error in the indicated temperatures resulting from electrolytic action.

Liquids and Corrosion Inhibitors

The liquids used in the tests were AN-E-2 othylene glycol (specified on a weight basis as 94.5 percent ethylene glycol, 2.5 percent triethanolamine phosphate, and 3 percent water), water, nominal (by volume) 70-30 and 30-70 glycol-water mixtures and commercial butanol (n-butyl alcohol). The glycol concentration in AN-E-2 ethylene glycol and the more aqueous glycol mixtures was determined from the specific gravity of samples taken at intervals throughout the tests.

A corrosion inhibitor, sodium chromate, was used in prelimiinary tests conducted with water. This practice was discontinued
before the final tests, however, because the sodium chromate was
believed to be affecting the liquid thermocouple calibrations. In
preliminary tests conducted with the glycol-water solutions, inconsistencies appeared in the results after very short periods of
operation. These inconsistencies were probably due to fouling of
the inside tube-wall surface even though the tube was thoroughly
cleaned with a fine-grade steel wool before every test series. As
a corrective measure, 0.2 percent by volume of NaMBT (sodium
mercaptobenzothiazole) was added to the AN-E-2 ethylene glycol and
the other glycol-water mixtures in the final tests. A corrosion
inhibitor was not used in the tests conducted with butanol.

PRELIMINARY TESTS

Various preliminary tests were conducted in order to check the accuracy of the heater-tube instrumentation. A detailed discussion of these tests is presented in appendix A. The results of the preliminary investigation indicated that: (a) the electric currents and magnetic fields in and around the tube did not introduce any noticeable error in the tube-wall thermocouple readings; (b) end losses affected the tube-wall temperature distribution at the end sections but had little effect on the temperature distribution of the central 12 inches; and (c) the electrical resistance of the tube per inch length as calculated from the ammeter and the voltmeter readings and the length of the tube could be used for power-input computations.

FINAL TESTS AND CALCULATIONS

Final Tests

Final tests were conducted to obtain forced-convection heattransfer coefficients for the various liquids over the following ranges of conditions:

Average liquid temperature, OF 100 to 250
Liquid-flow rate, pounds per second 0.17 to 2.50
Reynolds number
Heat flux, Btu per second per square foot 4 to 36

Each factor was independently varied while maintaining the other factors approximately constant. The tests were repeated at several different values of the constant factors. Most of the tests were conducted at approximately constant absolute liquid pressures of 53 to 70 pounds per square inch. In a few of the tests, however, each run was made at two different pressures in order to determine the effect of pressure on the heat-transfer coefficients. In all the tests a high enough pressure was maintained to obtain nonboiling conditions.

Calculations

The following symbols will be used in the calculations:

- A₁ inside area of test section of tube, (sq ft)
- Am mean area of test section of tube perpendicular to heat flow, (sq ft)
- c specific heat of liquid, (Btu)/(lb)(OF)
- D inside diameter of tube, (ft)
- E potential drop in test section of tube, (volts)
- G mass rate of liquid flow, (lb)/(sec)(sq ft)
- h heat-transfer coefficient, (Btu)/(sec)(sq ft)(OF)
- I tube current, (amperes)
- k thermal conductivity of liquid, (Btu)/(sec)(sq ft)(OF/ft)
- m, n exponents, experimentally determined
- p absolute liquid pressure, (lb)/(sq in.)
- q rate of heat input to test section of tube. (Btu)/(ssc)
- q' rate of heat input to entire tube length, (Btu)/(sec)
- qr rate of heat rejected to liquid, (Btu)/(sec)
- R electrical resistance of test section of tube, (ohms)

```
r
     electrical resistance of tube per inch length, (ohms)/(in,)
t
     average liquid temperature, (OF)
     average inside-wall temperature of test section of tube, (OF)
     average outside-wall temperature of test section of tube,
to
       (呼)
ta
     temperature of outside-surface of tube insulation, (OF)
Δt
     temperature rise of liquid in flowing through tube, (OF)
W
     liquid-flow rate, (lb)/(sec)
I
     thickness of tube wall, (ft)
     inside radius of tube, (ft)
Уi
     outside radius of tube, (ft)
Jo
     dimensionless constant
     constant, 0.000948, (Btu/sec)/(watt)
β
θ
     time. (sec)
     absolute viscosity of liquid. (lb)/(ft)(sec)
μ
     resistivity of tube, (ohms)(sq ft)/(ft)
cu/k Prandtl number
```

DG/μ Reynolds number hD/k Nusselt number

h/cG Stanton number

Average temperatures. - Average liquid temperatures t were taken as the arithmetic mean of the liquid-bulk temperatures measured at the entrance and exit ends of the tube. Average outside-tube-wall temperatures to were taken as the arithmetic average of the temperatures indicated by the 13 thermocouples spot-welded on the central 12 inches of the tube; that is, the test section of the tube was considered to consist of the central 12 inches in order to reduce the possibility of introducing errors in the final results

owing to the effect of end losses on the tube-wall temperature distribution at the end sections. (See appendix A.) Average inside-tube-wall temperatures t_i were calculated using the following relation, which is derived in appendix B:

$$q = \frac{k_s A_m}{x} \frac{(t_o - t_1)}{0.5}$$

where A_m is equal to 0.123 square foot and k_s is obtained from figure 4, prepared from references 4 and 5, at the value of the average outside-tube-wall temperature t_0 .

Power input and heat-transfer coefficients. - The power input to the tube q was calculated using the I^2R law where the total electrical resistance R is equal to the product of the resistance of the tube per inch length r and the length of the test section considered (12 in.). Figure 5 shows r as a function of temperature as determined in the check tests. Values of r were obtained at the value of the average outside-tube-wall temperature t_0 .

Heat-transfer coefficients h were calculated as follows:

$$h = \frac{q}{A_1 (t_1 - t)}$$

where A₁ is equal to 0.115 square foot.

Heat rejections and physical properties. - The total heat rejected to the liquid based on the full-length tube was calculated as follows:

$$q_r = Wc \Delta t$$

where Δt , the temperature rise of the liquid, was obtained from the differentially connected thermopiles.

The specific heat c, the thermal conductivity k, and the absolute viscosity μ of the liquids were determined at the value of the average liquid temperature t. Figures 6 and 7 (data from reference 2) and figure 8 (data from references 6, 7, and 8) show the physical property values of water, aqueous ethylene glycol solutions, and butanol, respectively, as a function of temperature.

The physical properties of the glycol-water mixtures were evaluated by assuming that the corrosion inhibitors in the solutions were approximately equal to an equivalent amount of ethylene glycol.

For example, the properties of AN-E-2 ethylene glycol were evaluated as for a nominal (by volume) 97-3 glycol-water mixture; hence, corrections were not made for the small effects of the corrosion inhibitors on the individual physical-property values. The errors introduced in the final results by making this assumption were relatively small.

RESULTS AND DISCUSSION

. Summary of Data

A summary of data and results for all of the tests except preliminary and check tests is presented in table I. The values presented for the heat rejected to the liquid represents the total heat rejected on the basis of the full-length tube (22.75 in.). The total heat rejected to the liquids is usually lower than the total electrical heat input. The maximum deviation is less than 10 percent in most cases. The heat loss through the thermal insulation on the tube was estimated to be less than 1 percent of the heat input and the remaining portion of the total heat loss is attributed to end losses through the copper adapters and busses. At the central 12-inch test section, however, heat-input measurements should be accurate measures of heat transfer inasmuch as the effect of end loss on this portion of the tube is negligible. (See appendix A.)

Individual Heat-Transfer Coefficients

The variation of the heat-transfer coefficient with rate of heat input is shown in figure 9 from the results of tests conducted with water at an average liquid temperature of approximately 150° F, at a liquid pressure of 65 pounds per square inch absolute, and at a liquid-flow rate of 0.20 pound per second. The heat-transfer coefficients remained approximately constant for variations in the power supplied to the tube. This constant relation is, in effect, a precision check of the entire setup inasmuch as a constant liquid-flow rate and average liquid temperature (hence physical properties) predicates constant heat-transfer coefficients. (See equation (1).)

The increase of heat-transfer coefficients with average liquid temperature for water and each of the glycol-water mixtures tested is shown in figure 10. The data were obtained at approximately constant liquid-flow rates of 0.67 pound per second and different constant heat inputs. In the tests conducted with the glycol-water mixtures, a constant liquid pressure of 68 pounds per square inch absolute was maintained, whereas in the tests conducted with water

each run was made at liquid pressures of 15 and 68 pounds per square inch absolute. In the water data, no appreciable effect of pressure was found. The change of the heat-transfer coefficients per degree Fahrenheit change in average liquid temperature and the value of the heat-transfer coefficients at 150° F and 200° F as obtained from figure 10 are listed in the following table:

Coolant glycol-	Slope	Heat-transfer coefficient, h (Btu)/(sec)(sq ft)(OF)								
water (percent by volume)		Average liqu	id temperature F)							
		150	200							
0-100 30-70 70-30 97-3	0.16 .18 .13	0.75 .52 .24 .16	0.83 .62 .30 .22							

The advantage in cooling performance of water and the more aqueous glycol solutions over AN-E-2 ethylene glycol is shown in the previous table. At an average liquid temperature of 200° F, the heat-transfer coefficients obtained using water, nominal (by volume) 30-70 and 70-30, glycol-water mixtures are approximately 3.8, 2.8, and 1.4 times higher, respectively, than the heat-transfer coefficients obtained using AN-E-2 ethylene glycol.

Correlation of Heat-Transfer Coefficients

The heat-transfer coefficients were correlated using the familiar Nusselt relation (reference 1, p. 164):

$$\frac{hD}{k} = \alpha \left(\frac{c\mu}{k}\right)^n \left(\frac{DG}{\mu}\right)^m \tag{1}$$

In order to determine the exponent of the Prandtl number in equation (1) two plots were made. The first plot (fig. ll(a)) shows the Nusselt number plotted on logarithmic coordinates against the Reynolds number as obtained from variable liquid-flow-rate tests conducted with water and the glycol-water solutions at various constant values of average liquid temperature (hence Prandtl number), power input, and liquid pressure. A family of approximately parallel lines was obtained.

From figure 11(a) at a value of the Reynolds number equal to 50,000, the values of the Nusselt number are cross-plotted on logarithmic coordinates against the Prandtl number in figure 11(b). The slope of the resulting line is approximately 0.4. This value, which is equal to the exponent of the Prandtl number (equation (1)), is in agreement with that found by investigators using various other liquids (reference 1, p. 167).

The first correlation plot of the heat-transfer coefficients presented in table I is shown in figure 12 in a logarithmic plot of $\frac{hD}{k} / \left(\frac{c\mu}{k}\right)^{0.4}$ against DG/ μ . The results for AN-E-2 ethylene glycol and the more aqueous glycol mixtures correlate well with the water and the butanol data. The value of the slope of the resulting line through the data is approximately 0.73. Although 0.8 is generally recommended as the exponent of the Reynolds number in equation (1), investigators using other liquids have found values of the exponent from about 0.7 to 0.8 (reference 1, pp. 178-181). The equation of the line through the data is as follows:

$$\frac{hD}{k} = 0.048 \left(\frac{c\mu}{k}\right)^{0.4} \left(\frac{DG}{\mu}\right)^{0.75} \tag{2}$$

The average scatter from this equation is approximately ±10 percent with a few of the points, especially those below a Reynolds number of 10,000, deviating slightly more than 10 percent.

The second correlation plot of the heat-transfer coefficients involving the Stanton number is shown in figure 13 in a logarithmic

plot of $\left(\frac{h}{cG}\right)\left(\frac{c\mu}{k}\right)^{0.6}$ against DG/ μ . This second method of correlating forced-convection heat-transfer data has the advantage of illustrating better than the first correlation plot trends in the neighborhood of the transition region. The effect of the transition is shown in figure 13 by the curvature of the data below a Reynolds number of approximately 10,000. The equation of the line for the data of Reynolds numbers greater than 10,000, which corresponds to equation (2), is:

$$\left(\frac{h}{\alpha G}\right) \left(\frac{c\mu}{k}\right)^{0.6} = 0.048 \left(\frac{DG}{\mu}\right)^{-0.27}$$

The average scatter from this equation is approximately ±10 percent with slightly larger deviations for some of the points.

SUMMARY OF RESULTS

The results of heat-transfer tests conducted with AN-E-2 ethylene glycol, nominal (by volume) 70 percent-30 percent and 30 percent-70 percent glycol-water solutions, water, and commercial butanol under turbulent flow conditions in an electrically heated tube showed that:

- 1. At a liquid-flow rate of 0.67 pound per second (Reynolds number, 14,500 to 112,500) and at an average liquid temperature of 200°F, the heat-transfer coefficients obtained using water, nominal (by volume) 30 percent-70 percent and 70 percent-30 percent glycolwater mixtures are approximately 3.8, 2.8, and 1.4 times higher, respectively, than the heat-transfer coefficients obtained using AN-E-2 ethylene glycol.
- 2. The heat-transfer coefficients of the coolants tested were satisfactorily correlated using the following equation:

$$\frac{hD}{k} = 0.048 \left(\frac{c\mu}{k}\right)^{0.4} \left(\frac{DG}{\mu}\right)^{0.73}$$

where h is heat-transfer coefficient; D is inside diameter of tube; k is thermal conductivity of liquid; c is specific heat of liquid; μ is absolute viscosity of liquid; and G is mass rate of liquid flow. In the evaluation of this equation, the physical properties used for the aqueous ethylene glycol solutions and water were those compiled by C. S. Cragoe (National Bureau of Standards) for the Coordinating Research Council.

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APPENDIX A

PRELIMINARY TESTS -

Various preliminary tests were conducted in order to check the accuracy of the heater-tube instrumentation.

Validity of measurements of outside-tube-wall temperatures. In view of the possible existence of errors in tube-wall-temperature
readings because of the electric currents and the magnetic fields in
and around the electrically heated tube, cooling-rate tests were conducted to check the validity of the tube-wall temperatures obtained.
These tests were conducted by supplying a small amount of power to
the tube devoid of any liquid until some predetermined temperature
was reached. At that point the power supply was cut off and the
indications of the thermocouples were recorded at intervals of
5 seconds for approximately 1 minute so that by extrapolation to
zero time, the initial temperature could be obtained.

The results of these tests are shown in figure 14 in a semi-logarithmic plot of t_0 - t_8 against θ . Linear relations were obtained from the results for the central thermocouples with slight deviations from linear relations obtained from the results for the end thermocouples. In an ideal case, straight lines would have been obtained for all of the results. The extrapolated temperatures at zero time, however, did check all of the initial tube-wall temperatures that were recorded when power was being supplied to the tube, thus indicating no appreciable error in the tube-wall-temperature readings.

Tube-wall temperature distribution. - In order to obtain an indication of: (a) the temperature distribution along the length of the tube and (b) the effect of end losses on the tube-wall temperatures, a test was conducted while the tube was dry wherein the heat input was set equal to the heat losses. After equilibrium was maintained for approximately 1/2 hour, all the thermocouple readings were recorded. The results of this test are shown in figure 15 where each tube-wall temperature is plotted with respect to the corresponding thermocouple location. The thermocouples located 9 and 10 inches from the entrance of the tube were inoperative when this test was conducted. The temperatures indicated by the other thermocouples located on the central 12 inches of the tube agreed within approximately ±8° F but on both sides of this central section the temperatures decreased rapidly. The resulting temperature distribution is, of course, greatly exaggerated when compared with the temperature distribution obtained under actual operating conditions.

an example of which is shown in figure 15. It is evident, however, that the temperatures along the central 12 inches of the tube are not appreciably affected by the end losses in either case. Therefore, by taking the central 12 inches of the tube as the test section, the effect of end losses is reduced to a minimum.

Measurements of electrical resistance of the tube. - The power input to any section of the tube may, of course, be calculated by the I²R law if the electrical resistance of the tube is known. The resistance of the tube per inch length was therefore obtained over a range of operating temperatures from the ammeter and the voltmeter readings and the length of the tube after the power factor of the tube was checked with an oscillograph and found to be unity. The results of the foregoing computations are shown in figure 5 where the resistance of the tube per inch length is shown as a function of temperature. The results of resistance measurements made with a Kelvin bridge using a sample tube of the same material agreed within approximately 1.5 percent with the alternating-current results.

APPENDIX B

CALCULATION OF INSIDE-TUBE-WALL TEMPERATURES

In order to evaluate the heat-transfer coefficient between the inside-tube-wall surface and the liquid, a relation for calculating the inside-tube-wall temperature from the measured outside-tube-wall temperature was obtained.

It is known that if all the heat passes the full thickness x of the tube wall, then the temperature drop through the wall could be calculated from the following familiar equation:

$$q = \frac{k_{B} A_{m}}{T} (t_{O} - t_{1}) \qquad (1)$$

In view of the fact that the heat is produced by an electric current flowing through the tube and, hence all of the heat going to the liquid does not pass the full thickness of the tube, equation (1) is not valid for this application.

A relation similar to equation (1), however, giving the actual temperature drop through the tube wall can be obtained as follows:

Assuming that: (a) the heat is produced in the tube uniformly across the tube-wall thickness; and (b) the heat flow is in only one direction (toward the liquid), for a section of unit length the heat generated from the outer radius y_0 to any other radius y is as follows:

$$q = \beta \frac{\pi E^2}{\rho} (y_0^2 - y^2)$$
 (2)

and the heat conducted is:

$$q = 2\pi y k_s \frac{dt}{dv}$$
 (3)

Combination of equations (2) and (3) results in the following:

$$\beta \frac{\pi E^2}{\rho} (y_0^2 - y^2) = 2\pi y k_s \frac{dt}{dy}$$

or separating the variables and rewriting:

$$dt = \beta \frac{\pi E^2}{2\pi \rho k_B} \left(\frac{y_0^2 - y^2}{y} \right) dy$$

Integrating between the limits of y_i and y_o and t_i and t_o:

$$t_{o} - t_{1} = \beta \frac{\pi E^{2} y_{o}^{2} \log_{e} \left(\frac{y_{o}}{y_{1}}\right)}{2\pi \rho k_{g}} - \beta \frac{\pi E^{2} (y_{o}^{2} - y_{1}^{2})}{4\pi \rho k_{g}}$$
(4)

But the total heat produced can be expressed as

$$q = \beta \frac{\pi E^2}{\rho} (y_0^2 - y_1^2)$$

and substituting from this expression into equation (4) results in the following:

$$t_0 - t_1 = \frac{q y_0^2 \log_e(\frac{y_0}{y_1})}{2\pi k_g (y_0^2 - y_1^2)} - \frac{q}{4\pi k_g}$$

or rewriting

$$q = \frac{2\pi k_{s} (t_{o} - t_{i})}{\log_{e} \left(\frac{y_{o}}{y_{i}}\right) \left[\frac{y_{o}^{2}}{y_{o}^{2} - y_{i}^{2}} - \frac{1}{2 \log_{e} \left(\frac{y_{o}}{y_{i}}\right)}\right]}$$
(5)

Substituting the values of y_0 (0.0208 ft) and y_1 (0.0182 ft) in the bracketed term and noting that $\frac{2\pi}{\log_{\Theta}\left(\frac{y_0}{y_1}\right)}$ is equal to $\frac{A_m}{x}$

results in the following expression, which gives the theoretical temperature drop through the tube wall:

$$q = \frac{k_g A_m}{x} \left(\frac{t_0 - t_1}{0.525} \right)$$
 (6)

The constant 0.525 in equation (6) was rounded off to 0.5 for the actual calculations because it was felt that the simplifying assumptions used in the derivations did not justify additional significant places.

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Run	Tube	Heat	rate, Btu/sec		Liquid-	Liquid		Liquid	Average t	ube-wall	Heat-transfer		Reynolds		
	current	Inpu		Rejected	flow rate	tempera	ture	pressure	temperatu	re of	coefficient	number	number	number	number
	(amp)	Test section q (center 12 in.)	Full section	to liquid q _r	(lb/sec)	(°F) Average t	Riee Δt	p (lb/sq in. abso- lute)	center 12 of test s (°F	ection	h (Btu)/(sec) (sq ft)(°F)	cμ/kr	DG/µ	hD/k	h/cG
									Outside to	Incide t _i					•
				Test w	ith variab	le avera	ge li	quid tempe	rature; li	quid, wa	ter				
126 127	660 660	3.00 3.02	5.65 5.67	5.19 5.15	0.33	122.3 136.5	15.9 16.0		190 203	178 191	0.47 .49	3.5 3.1	31,000 34,700	167.2 170.3	0.00151
128 129	6 53 6 4 8	2.98 2.95	5.60 5.56	5.05 5.09	.32 .31	153.8 160.0	15.9 16.2	56	218 228	206 216	.50 .46	2.6 2.5	39,400 40,800	172.7 157.7	.00165 .00153
130 131 132	643 643	2.92 2.94	5.50 5.55	4.93 5.26	.31 .33	173.0 187.5	16.0 15.8	58	235 248	223 236	.51 .53	2.3 2.1	44,100 52,400	174.0 178.2	.00172 .00166
133	686 650	3.31 2.95	6.2 5 5.56	5.60 4.90	.32 .32	152.0 148.8	17.7 15.5		223 213	210 201	.50 .49	2.7 2.8	38,700 37,700	173.3 169.7	.00165 .00162
142 143	677 674	3.21 3.21	6.11 6.07	5.39 5.41	.18 .19	114.5 129.3	29.5 29.3	57	233 233	209 220	.30 .31	3.9 3.3	16,200 18,700	106.2 108.4	.00171 .00175
144 145	667 670	3.16 3.21	5.97 6.06	5.35 5.53	.18 .19	143.8 157.5	29.4 29.6		2 44 261	231 248	.32 .31	2.9	20,800 2 5, 900	109.1 106.1	.00184 .00173
146 147	665 660	3.19 3.17	6.02 5.97	5.37 5.29	.18 .18	173.8 189.5	29.8 29.6	64	275 290	262 277	.31 .31	2.3 2.0	26,000 28,600	106.6 105.3	.00179 .00181
148 149	674 674	3.19 3.21	6.01	6.00 5.58	.20 .19	115.0	30.5	65	220 232	207 219	.30 .32	3.8 3.2	17,400 19,700	107.9 111.9	.00160
150 151	672 667	3.20 3.18	6.04 6.01	5.46 5.43	.19 .19	146.3	29.2	60	244 255	231 242	.33 .33	2.5	21,800 24,100	114.1	.00185
152 153 154	665 667 665	3.18 3.24 3.17	5.99 6.08 5.96	5.28 5.37 5.35	.18 .18 .19	172.8 190.0 174.0	29.3 29.4 28.4	62 63 60	270 290 2 64	257 277 25 1	.33 .32 .36	2.3 2.0 2.3	25,800 29,200 27,200	112.0 109.2 120.7	.00191 .00183
155 156	670 677	3.18 3.22	6.00	5.35 5.68	.19 .20	145.0	28.6		245 225	232 212	.32 .29	2.9	21,600	110.2	.00179
157 158	684 682	3.21 3.18	6.05	5.73 5.72	.58 .58	125.5	10.0	63	180 179	167 166	.69 .69	3.4 3.4	56,200 56,200	242.0	.00125
160	806 802	4.46 4.44	8.44 8.38	8.06 7.90	.58	103.8	14.0	63	182 194	163 176	.65 .68	4.4 3.7	45,600 52,400	235.0 241.8	.00118
162 163	797 799	4.41 4.47	8.33 8.43	7.85 7.94	.57 .87	129.3 144.3	13.7 13.9	63	203 217	185 199	.70 .72	3.3 2.9	58,000 65,900	243.7 248.5	.00128
164 165	797 78 5	4.47 4.36	8.41 8.23	7.91 7.59	.57 .56	158.5 173.3	13.8 13.6	63	229 240	211 222	.75 .77	2.5 2.3	73,600 80,200	257.1 262.4	.00136
166 167	773 775	4.25 4.31	8.06 8.11	7.43 7.53	.56 .56	182.8 198.0	13.3 13.4	66	247 262	230 245	.78 .80	2.1 1.9	85,500 94,600	264.0 269.2	.00145 .00149
168	775 778	4.28 4.28	8.09 8.12	7.68 7.66	.89 .88	196.5 183.5	8.6		249 238	232 221	1.05	1.9 2.1	149,400	352.9 343.3	.00123
170 171 172	785 794 797	4.33 4.41 4.40	8.25 8.36 8.34	7.82 7.88 8.03	.89 .89 .91	173.5 158.0 143.0	8.8 8.9 8.8	64	228 2 15 201	210 197	1.03 .99 .97	2.5	127,900	349.4 339.3	.00121
173 174	806 811	4.47	8.41 8.46	8.06 8.14	.90	125.5	8.9	63	185 165	183 166 146	.97 .95 .86	2.9 3.4 4.5	103,000 88,100 68.900	335.5 333.5 310.8	.00112 .00110
192 193	778	4.33	8.20 8.07	7.85 7.61	.20	118.5	40.1 38.9	68	260 263	243 246	.31 .31	3.7 3.4	17,900 19,300	108.2	.00166
194 195	773 768	4.31	8.15 8.10	7.73	.20	143.0	39.5		27 4 285	257 268	.33	2.9	22,200	113.9	.00176
196 197	761 761	4.22 4.23	8.03 8.06	6.93	.18	170.5 176.5	39.1 39.0	68	295 301	278 284	.34	2.3	24,900 25,900	116.1 115.8	.00200

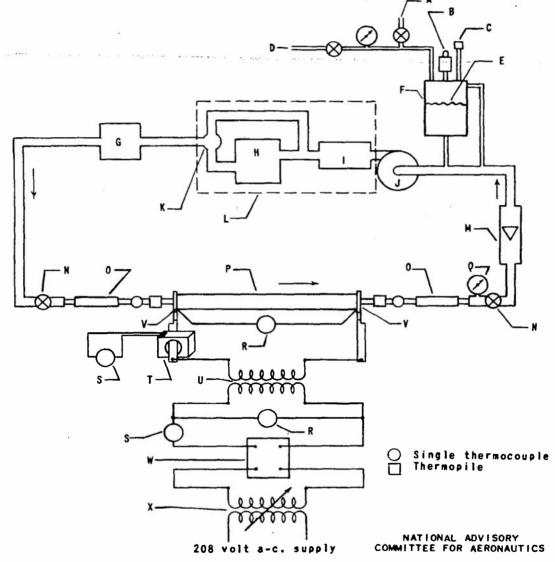
													TITEE PO		
Run	Tube current I (amp)	Heat Input Test section	Pull section	Rejected to liquid	Liquid- flow rate W (lb/sec)	(°F)		Liquid pressure p (1b/sq	temperatu center 12 of test s	re of inches ection	Heat-transfer coefficient h (Btu)/(sec)	Prandtl number cµ/k	Reynolds number DG/µ	Nusselt number hD/k	Stanton number h/cG
	` • •	(center 12 in.)	(22.75 in.)	q _r		Average t	At Δt	in. abso- lute)	Outside to	Inside	(sq ft)(°F)				
		·	Test	t with var	able aver	age liqu	id te r	perature;			oncluded	•		,	
309 310 311 312 313 314 333 334 335 336 339 366 367 368 369 370 371 372	782 782 782 773 763 742 749 768 773 794 751 761 768 780 780 785	4.35 4.39 4.39 4.25 4.25 4.25 4.01 4.20 4.15 4.18 4.28 4.03 4.05 4.11 4.09 4.12	8.27 8.28 8.31 8.15 8.06 8.07 7.45 7.50 7.77 7.80 7.72 7.68 7.72 7.68 7.74 7.87	7.53 7.57 7.66 7.56 7.55 6.98 8.24 7.92 7.98 8.06 8.11 8.05 7.37 7.79 7.81 7.81 8.00	0.20 .20 .20 .20 .18 .83 .84 .83 .84 .83 .67 .67 .67	94.3 119.3 130.3 144.3 156.5 176.3 247.9 223.5 197.8 173.3 149.8 123.5 94.9 197.7 197.3 172.3 171.9 146.4 121.7 98.9	38.7 38.9 39.3 38.7 39.8 9.9 9.6 9.7 9.6 9.8 11.5 11.5 11.7	69 69 69 69 69 70 70 70 70 70 70 68 15 68 68 68	246 258 266 273 284 300 295 273 251 229 206 183 160 256 235 234 213 190 171	228 240 248 256 267 283 279 257 234 212 189 166 142 240 240 218 196 173 153	0.28 .32 .33 .34 .35 1.14 1.05 1.01 .96 .93 .86 .79 .84 .84 .78 .78	4.9 3.7 3.3 2.9 2.2 1.4 1.6 2.3 2.7 3.5 4.9 1.9 1.9 2.3 2.3 2.3 4.9	13,900 18,000 19,900 22,600 24,800 25,700 187,400 165,100 139,700 119,800 100,700 79,300 59,200 96,200 96,200 94,900 78,300 64,000 50,300	104.4 112.0 114.0 116.2 114.7 117.9 377.4 349.2 338.6 323.9 318.9 304.2 290.3 280.8 281.3 266.0 252.8 247.2	0.00150 .00172 .00181 .00182 .00208 .00142 .00130 .00127 .00126 .00109 .00100 .00131 .00130 .00122 .00124 .00111
				Tes	t with var	riable li	quid-	flow rate	liquid,	water	.	.	L	l	<u> </u>
119 120 121 122 123 124 125 134 137 138 139 140 141 175 176 177 178 179 180 181	648 650 653 663 660 655 667 674 679 684 682 761 768 778 785 787	2.93 2.94 2.94 2.99 2.99 2.99 2.94 3.18 3.21 3.24 3.23 3.21 4.22 4.22 4.22 4.22 4.23 4.34 4.36	5.52 5.53 5.53 5.61 5.62 5.53 5.98 6.03 6.08 6.09 6.07 5.92 8.04 8.03 8.12 8.23 8.23	4.89 4.86 4.78 4.67 4.57 4.34 5.57 5.68 5.71 5.79 5.79 5.79 7.20 7.54 7.81 7.89	0.32 .48 .65 .81 .128 .19 .32 .49 .65 .82 .98 .82 .18 .19 .31 .48 .89 .11		15.4 10.1 7.4 5.7 4.7 4.0 3.4 29.1 17.3 11.7 8.8 7.0 5.9 7.1 30.0 36.9 12.0 8.9	57 59 57 57 60 57 56 53 59 55 53 53 62 62 61 61 61	214 200 194 189 185 184 181 243 219 205 198 199 193 244 296 267 251 241 233 231 226	202 168 182 177 173 172 169 230 206 192 185 177 180 231 279 250 234 224 215 213	0.51 .68 .83 .98 1.10 1.20 1.31 .34 .49 .66 .80 .93 1.05 .90 .33 .37 .52 .68 .82 1.02	2.7 2.7 2.7 2.7 2.8 2.7 2.8 2.7 2.8 2.7 2.8 2.2 2.2 2.2 2.2	38,700 57,600 78,500 98,600 116,600 137,500 152,400 22,600 38,200 58,600 75,200 97,500 117,000 97,000 47,000 47,000 71,600 95,300 132,200 143,700 168,400	175.8 232.2 284.0 336.3 379.7 414.1 117.1 117.1 1274.9 319.9 360.1 309.0 114.3 124.6 176.7 231.1 278.1 345.7 356.0 398.5	0.00168 .00148 .00133 .00126 .00117 .00110 .00169 .00159 .00142 .00128 .00119 .00112 .00115 .00114 .00173 .00147 .00147 .00147

	Tube	Heat	rate, Btu/eec		Liquid-	Liquid		Liquid			Heat-transfer	Prandtl	Reynolds	Nusselt	
	current I	Inpu	t	Rejected	flow rate	tempera (°F)	ture	pressure	temperatu		coefficient h	number cµ/k	number DG/µ	number hD/k	number h/cG
	(amp)	Test section q (center 12 in.)	Full section q' (22.75 in.)	to liquid	(lb/sec)	Average	Rise At	(lb/sq in. abso- lute)	of test s	ection	(Btu)/(zec) (eq ft)(°F)	сµ/к	. 10/д	nu/,k	n/eu
	:							,	Outside to	Inside t _i					
				Test wit	th variable	liquid	-flow	rate; liq	uid, water	- Concl	uded				
182	790	4.38	8.29	7 .9 2	1.29	178.5	6.1	61	226	208	1.28	2.2	191,400	431.4	0.00103
183	775	4.30	8.13	7.64	.20	127.1	37.7	62	260	243	.32	3.4	20,100	113.9	.00165
184 185	785 794	4.35 4.39	8.18 8.28	7.72 7.91	.33 .50	128.5 128.5	23.6 15.8	62 64	228 208	210 190	.46 .63	3.3	32,900	163.0 219.7	.00146
186	797	4.41	8.31	7.98	.5B	127.5	13.9	64	202	184	.68	3.4	50,200 57,400	238.9	.00132
187	797	4.40	8.31	8.06	.66	128.3	12.2	64	198	180	74	3.3	66,000	260.4	.00117
188	802	4.43	8. 37	B.13	.83	127.3	9.8	64	190	172	.87	3.4	82,200	305.7	.00110
189	809	4.50	8.48	8.33	.99	127.3	8.5	64	186	167	.98	3.4	97,700	342.2	.00104
190	814	4.54	9.18	8.36	1.15	127.0	7.3	63	183	164	1.08	3.4	113,800	380.7	.00098
191	814	4.54	8.55	8.43	1.31	129.3	6.5	63	181	162	1.20	3.3	132,100	420.8	.00096
340	749	4.02	7.49	7.78	.21	148.3	36.8	69	265	249	.35	2.8	25,100	120.6	.00173
341	749	4.02	7.48	7.77	.21	147.5	36.4	69	263	247	.35	2.8	25,200	121.6	.00171
342	758	4.07	7.58	8.20	.33	147.3	24.9	69	237	220	.49	2.8	38,800	167.1	.00158
343	761	4.06	7.60	8.66	.50	150.0	17.3	69	222	205	.64	2.7	60,200	221.1	.00134
344	763	4.06	7.60	B.80	.66	149.5	13.3	69	212	195	.77	2.8	78,900	264.7	.00122
345	773	4.15	7.68 7.85	8.63 8.53	.82 1.00	148.5	10.5 8.5	69 69	206 201	189 183	.91 1.05	2.8	97,500	312.6 362.7	.00115
346 347	782 778	4.25 4.18	7.75	8.81	1.15	148.0	7.7	69	197	180	1.15	2.8	118,400	395.5	.00105
348	780	4.21	7.77	8.82	1.29	149.8	6.8	69	197	180	1.25	2.7	154,800	431.4	.00101
350	734	3.90	7.29	7.49	.17	150.3	43.2	- 68	280	264	.30	2.7	20,900	102.9	.00181
351	737	3.93	7.36	7.68	.17	148.5	44.4	49	281	265	.29	2.8	20,600	101.0	.00175
352	751	3.99	7.47	7.55	.34	150.3	22.2	68	236	220	.50	2.7	40,900	171.3	.00154
353	751	3.99	7.46	7.63	.34	150.3	22.5	45	236	220	- 50	2.7	40,900	171.3	.00154
854	756	4.03	7.55	7.59	.51	147.3	15.0	68	218	201	.65	2.8	59,400	223.8	.00134
355	758	4.03	7.55	7.61	.50	148.0	15.1	33	219	202	.65	2.8	59,400	222.6	.00135
356	754	3.95	7.38	7.30	.68	148.5	10.8	68	207	191	.82	2.8	79,900	280.7	.00127
357	754	3.95	7.39	7.33	.67	149.3	10.9	15	209	193	.81	2.8	80,200	276.8	.00126
358 359	758 758	3.99 3.99	7.45 7.48	7.38 7.36	.83 .83	147.5	8.9 8.9	68 15	202 204	185 187	.91	2.8	97,800 98,900	314.3 315.7	.00114
360	763	4.03	7.52	7.15	1.00	146.8	7.2	68	199	182	1.00	2.8	116,200	344.8	.00105
361	763	4.03	7.54	7.23	1.00	146.8	7.3	15	199	182	1.00	2.8	116,200	343.9	.00105
362	763	4.03	7.55	7.24	1.16	149.3	6.3	68	196	180	1.16	2.8	138,500	398.0	.00105
363	761	4.00	7.49	7.29	1.15	148.3	6.3	15	196	180	1.12	2.8	136,100	384.5	.00102
364	761	4.00	7.50	7.16	1.31	148.8	5.5	68	194	178	1.21	2.8	155,000	417.2	.00097
365	763	4.02	7.52	7.17	1.31	148.5	5.5	15	194	177	1.22	2.8	155,200	419.8	.00098
495	751	3.68	7.32	6.20	2.48	150.5	2.5	6 8	184	168	1.92	2.7	299,600	661.1	.00061
496	754	3.90	7.36	6.60	2.20	150.5	3.0	68	186	170	1.81	2.7	265,800	623.3	.00086
497	749	3.86	7.29	6.53	1.92	151.0	3.4	68	1 8 8	172	1.64	2.7	232,900	564.9	.00089
498	749	3.86	7.28	6.42	1.69	149.5	3.8	68	189	173	1.47	2.7	201,800	504.3	.00091
499	746	3.85	7.22	6.26	1.36	152.5	4.6	68	192	176	1.41	2.7	166,600	484.2	.00108
500	751	3.90	7.31	6.51	1.21	154.0	5.4	68	191	175	1.31	2.6	149,700	449.6	.00113
501	751	3.91	7.33	6.80	1.00	151.0	6.8	68 60	198	182	1.10	2.7	121,300	377.0	.00115
502	756	3.98	7.42	6.96	.75	150.5	9.3	6 8	207	191	.87	2.7	90,400	299.1	.00121

707				
159 168	040 046 048 051 047 043 043	057 046 045 050 033	033 036	041
.00	.00 .00 .00 .00	.00	.00	

				_								00/////////	EE FOR	ACITOTIAO	1105
	Tube current		rate, Btu/eec		Liquid- flow rate	Liquid		Liquid		tube-wall	Heat-transfer coefficient	Prandtl number	Reynolds	Nusselt number	nımhan
	I (amp)	Test eection q (center 12 in.)	Full section	Rejected to liquid ^q r	(lb/sec)	tempera (°F) Averege		pressure p (lb/sq in abso-	temperaticenter 12 of test :	2 inches section	h (Btu)/(sec) (sq ft)(°P)	cμ/k	DG/µ	hD/k	h/cG
		(00.001 12 1)	(22110 2111)					lute)	Outside t _o	Inside t _i				<u> </u>	
			· · · · · · · · · · · · · · · · · · ·		Test with	variabl	e hea	t input; 1	iquid, wat	ter		·····			
323 324 325 326 327	331 492 614 739 773	0.75 1.68 2.67 3.94 4.33	1.41 3.17 5.03 7.47 8.15	1.13 2.72 4.51 6.81 7.51	0.20 .20 .20 .20 .20	150.8 150.8 149.0 150.0 149.3	5.7 13.8 22.9 34.4 37.9	69 69 69 69	175 205 236 274 284	172 198 226 258 267	0.31 .31 .30 .32 .32	2.7 2.7 2.8 2.7 2.8	23,800 24,000 23,400 23,900 23,800	105.6 106.3 104.7 109.6 110.5	0.00165 .00163 .00159 .00168
			Test	with varia	ble averag	e liquid	temp	erature; 1	iquid, AN-	E-2 ethy	lene glycol				
481 482 483 484 515 516 517 518 519 520 521	394 396 394 389 298 298 298 298 298 298	1.08 1.10 1.09 1.08 .62 .62 .62 .61	2.04 2.06 2.05 2.03 1.13 1.14 1.14 1.14 1.14	1.81 1.80 1.71 1.74 .84 .98 1.02 1.02 1.03 1.01	0.67 .67 .67 .67 .97 .96 .95 .96 .97	123.1 150.0 171.5 198.3 199.5 180.0 171.5 163.0 153.0 140.5	4.4 4.3 4.0 4.0 1.3 1.6 1.7 1.7 1.7	68 68 68 68 68 68 68 68 68	209 214 227 249 220 213 204 196 188 180	205 210 223 245 217 210 201 193 185 177 167	0.12 .16 .19 .20 .31 .28 .25 .25 .24	59.9 41.8 32.3 25.5 25.2 26.9 29.8 32.4 35.6 40.1 47.0	5,000 7,700 10,500 14,500 23,100 21,000 16,200 14,500 12,400 10,300	99.2 143.6 172.9 197.5 291.0 257.6 228.8 221.8 209.5 185.4 172.6	0.00031 .00040 .00046 .00048 .00051 .00047 .00043 .00043 .00041 .00037
522	302	.62	1.15	1.08	.94	132.0	1.9	68	165	162	.18	53.0	8,800	153.2	.00032
							low r	ate; liqui				 	· · · · · · · ·		
476 477 478 479 480 503 504 505 506 507 508 509 510 511 512	389 384 391 394 389 278 278 278 278 278 276 276 276 276 276	1.07 1.08 1.09 1.07 .52 .53 .53 .53 .53 .53 .53 .53 .53 .53 .53	2.03 2.04 2.04 2.05 2.05 2.03 .94 .95 .95 .95 .95 .95 .94 .94 .94	1.74 1.72 1.73 1.69 1.73 1.32 1.18 1.16 1.08 1.05 .91 .84 .89 .87	0.67 .33 1.00 1.25 .67 2.51 2.25 1.99 1.70 1.42 1.14 1.28 1.09 .89 .71	199.5 200.5 197.9 197.7 150.5 150.5 150.5 151.0 151.0 151.5 152.0 151.5	4.00 7.8 2.6 2.1 4.00 .9 1.00 1.2 1.3 1.5 1.9 2.6	68 68 68 68 68	246 280 235 230 247 162 162 164 166 167 171 168 168 171 174 181	242 276 231 226 243 160 160 162 164 165 166 166 166 172 179	0.22 .12 .29 .35 .21 .49 .44 .43 .35 .30 .26 .33 .29 .25 .20	25.2 25.1 25.7 25.8 25.5 41.3 41.3 41.3 41.9 41.1 40.9 41.1 40.7 40.9	14,900 7,500 21,900 26,900 14,600 31,500 27,800 21,400 17,600 14,400 15,900 13,800 9,100 6,200	215.0 120.4 282.5 335.9 206.3 426.2 379.0 369.5 262.4 227.7 285.6 247.4 216.6 174.2 150.9 101.2	0.00052 .00057 .00046 .00045 .00033 .00033 .00036 .00038 .00038 .00038 .00038 .00038 .00038
		Test wit	h variable av	erage liqu	id tempera	ture; li	quid,	nominal (by volume) 70 perc	ent-30 percent	glycol-	water		
420 421 422 423 424 425	322 326 326 326 329 331	0.74 .73 .74 .75 .73	1.38 1.37 1.39 1.38 1.39 1.38	1.04 1.20 1.23 1.22 1.29 1.24	0.67 .67 .67 .67 .67	224.8 199.3 174.8 149.0 124.8 100.0	2.0 2.3 2.4 2.5 2.7 2.6	68 68 68 68 68	247 224 201 179 159 140	244 221 198 176 156 137	0.33 .30 .28 .24 .21	9.8 11.8 14.6 18.7 24.8 34.5	36,800 29,500 23,000 17,200 12,500 8,600	238.2 209.7 190.3 160.8 137.4 112.3	0.00065 .00061 .00057 .00051 .00045

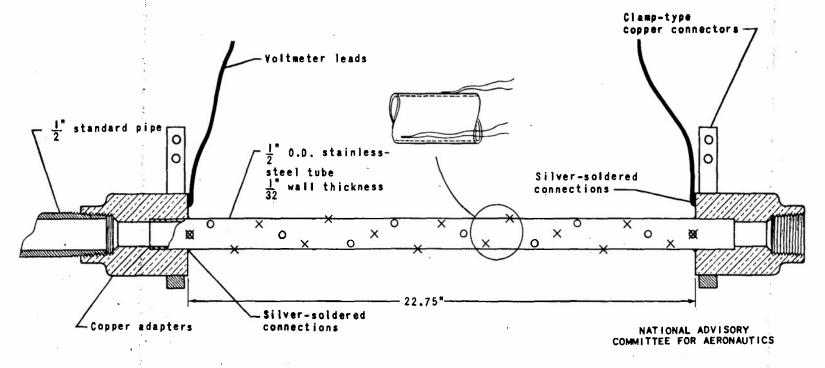
	Tube	Heat	rate, Btu/sec		Liquid-	Liquid		Liquid	Average t		Heat-transfer		Raynolds		
	currant I (amp)	Inpu Test eection	Full section	Rejected to liquid qr	flow rate W (lb/sec)	tempera (°F) Average	,	prassure p (1b/sq	temperatu center 12 of test s	inches ection	(Btu)/(sac)	number cµ/k	number DG/µ	number hD/k	number h/cG
		(center 12 in.)	(22.75 in.)	•		t	Δt	in. abso- lute)	Outside	Inside	(sq ft)(°F)				
Ш				. 14 4 4 6	<u> </u>	L	L		t _o	t ₁			100		L
┝		Test	WICH VAPIABL	e liquia-i.	low rate;	liquia, i	nomina	I (DA AOT.	·	rcent-30	percent glyco				
412 413 414 415 416 417	324 326 3 26 326 326 329	0.73 .74 .73 .73 .73	1.38 1.38 1.38 1.38 1.38	1.18 1.20 1.26 1.23 1.26 1.27	0.25 .34 .50 .68 .83 1.01	150.9 150.0 149.4 148.9 149.5 149.8	6.2 4.8 3.4 2.5 2.0 1.7	68 68 68 68 68	209 195 183 176 173 170	206 192 180 173 170 167	0.12 .15 .21 .26 .31	18.3 18.5 18.6 18.7 18.6	6,700 8,800 12,900 17,300 21,600 26,200	78.2 103.6 143.0 177.1 211.7 261.3	0.00066 .00063 .00059 .00054 .00052
418 419	329 331	.74 .75	1.38 1.39	1.34	1.18	149.3	1.5	68 68	167 167	164 164	.43 .45	18.6	30,500 34,300	288.9	.00051
Test with variable average liquid temperature; liquid, nominal (by												10.00		- 0811	
457 458 459 460 461 462 463 464 465 466 467	626 619 622 622 626 629 631 626 619 622 629 636	2.72 2.76 2.76 2.72 2.72 2.72 2.71 2.73 2.73 2.76 2.73 2.72 2.72	5.15 5.07 5.07 5.06 5.07 5.06 5.07 5.06 5.07 5.06 5.06 5.01	4.73 4.75 4.73 4.73 4.61 4.59 4.62 4.82 4.71 4.71	0.67 .66 .67 .67 .67 .67 .67 .67 .67	150.5 220.8 197.5 173.5 144.0 121.8 100.0 149.5 222.8 197.3 171.8 123.5 96.8	7.8 7.7 7.6 7.7 7.6 7.9 7.8 7.9 7.9	68 68 68 68 68 68 68 68 68 68 68	203 268 249 226 202 183 165 204 271 248 226 184	192 257 238 215 191 172 154 193 260 237 215 173 153 193	0.58 .66 .60 .58 .51 .47 .44 .55 .65 .65	6.2 3.6 4.2 5.0 6.6 91.0 6.6 4.2 5.1 5.1 5.1 5.1	43,200 74,900 64,900 52,100 40,500 31,600 24,300 42,600 77,000 64,600 52,400 32,400 23,200 41,400	264.7 305.0 273.3 262.4 234.1 219.3 205.2 255.0 299.1 271.7 249.3 223.7 199.6	0.00100 .00112 .00101 .00098 .00082 .00077 .00095 .00101 .00094 .00084
470	626	2.73	5.11 5.06	4.79	.67 .66	146.8	8.0	····	165 204	٠	.43 .52	6.4	41,400	239.7	00000
440 441 442 443 444 445 446 447 471 472 473	614 617 622 622 626 629 629 629 629 629 619	2.70 2.70 2.71 2.72 2.73 2.73 2.73 2.72 2.73 2.72 2.73	5.05 5.04 5.03 5.04 5.06 5.07 5.05 5.07 5.06 5.07 5.06 5.07	4.79 4.79 4.69 4.86 4.79 4.76 4.85 4.91 4.82 5.13 4.80	0.25 .34 .50 .67 .84 1.00 1.17 1.33 .67	149.0 148.0	20.8 15.6 10.3 8.0 6.3 5.2 4.1 8.0 4.2 21.2	68 68 68 68 68 68 68 68 68 68 68	260 244 223 213 203 194 190 190 204 167 250	249 233 212 202 192 193 179 179 179 176 239	0.24 -28 -37 -45 -54 -68 -74 -83 -62 -93 -25	6.3 6.3 6.4 6.4 6.4 6.4 6.4	16,100 21,400 31,700 42,500 52,700 62,600 72,800 41,600 83,300 15,200	108.1 127.3 172.3 205.5 247.1 310.9 340.3 381.2 239.1 382.4 116.8	0.00109 .00395 .00085 .00077 .00074 .00078 .00073 .00071 .00090 .00072
			Tost	with vari	able aver	age liqui	ld tem	perature;	liquid, c	ommercial	butanol		·		
529 530 531 532 533	364 390 394 396 398	1.04 1.07 1.08 1.08 1.08	1.97 2.00 2.01 2.03 2.04	1.99 1.98 1.99 2.07 2.06	0.31 .81 .81 .80	202.5 182.5 160.5 144.0 121.0	3.2 3.3 3.5 3.8 4.0	68 68 68 68	236 220 202 188 170	232 216 198 183 165	0.31 .28 .25 .24 .21	11.7 13.8 16.6 19.1 23.4	70,300 56,300 44,700 37,600 28,800	430.1 386.9 350.4 328.4 293.7	0.00052 .00049 .00046 .00046 .00043
				Test with	variable	liquid-	low r	ato; liqu	id, commer	cial buta	nol				
523 524 525 526 526 527 528	413 422 422 422 425 425	1.21 1.25 1.24 1.24 1.25 1.24	2.27 2.32 2.33 2.32 2.34 2.34	2.29 2.32 2.36 2.35 2.34 2.41	0.35 .52 .74 .86 .99 1.16	151.0 152.3 154.0 153.5 153.0 152.0	9.5 4.6 4.5 3.5 5.0	68 68 68 68 68	239 218 204 199 194 190	234 213 199 194 189 185	0.13 .18 .24 .27 .30 .33	18.0 17.8 17.5 17.6 17.7 17.8	17,600 26,300 38,100 44,200 50,200 58,500	175.3 250.6 334.0 373.8 419.1 459.0	0.00056 .00053 .00049 .00047 .00046



- A Air bleed line
 B Pressure-relief valve
- C Filling cap
- D Compressed-air connection
- E Liquid level
- F. Expansion tank
- 6 Filter
- H Cooler
- ! Electric heater
- J Pump
- K Temperature regulator
- L Heating and cooling blending unit

- M Rotameter
- * Throttle valve
- O Electrical-insulating coupling
- P Heater tube
- Q Pressure gage
- R Voltmeter
- S Ammeter
- T 240:1 current transformer
- 20:1 power transformer
- y Clamp-type copper connectors
- W Voltage regulator
- X Autotransformer

Figure 1. - Schematic diagram of heater-tube setup and associated equipment.



X Front-side thermocouple O Rear-side thermocouple

Figure 2. - Details of heater-tube section showing thermocouple locations and electrical-system and liquid-system connections.

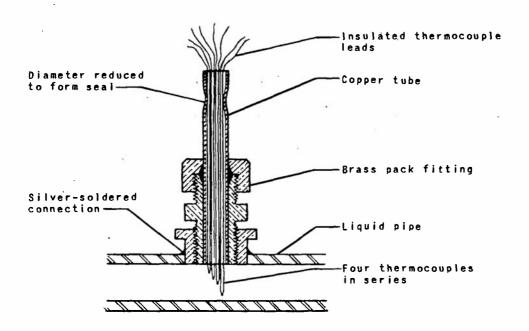


Figure 3. - Thermopile construction and installation in liquid pipes.

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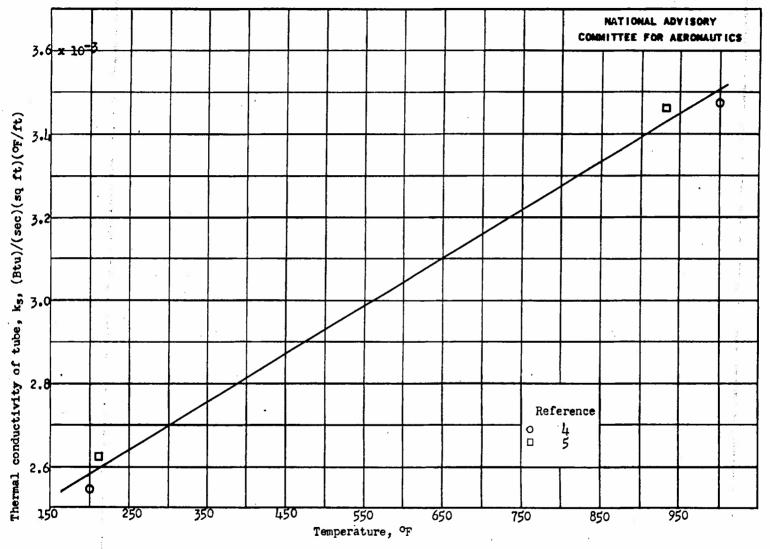


Figure 4.- Variation of thermal conductivity of 18:8 stainless steel with temperature. (Data from references 4 and 5.)

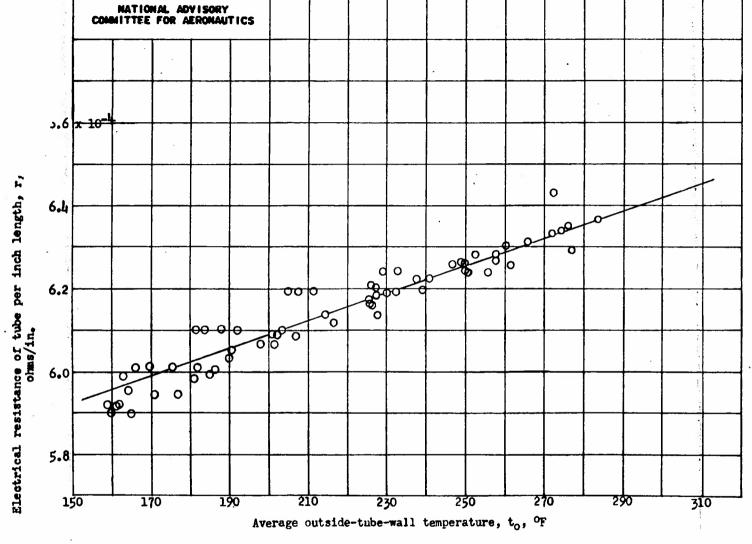


Figure 5.- Variation of electrical resistance of tube per inch length with average outside-tube-wall temperature as obtained from alternating-current measurements.

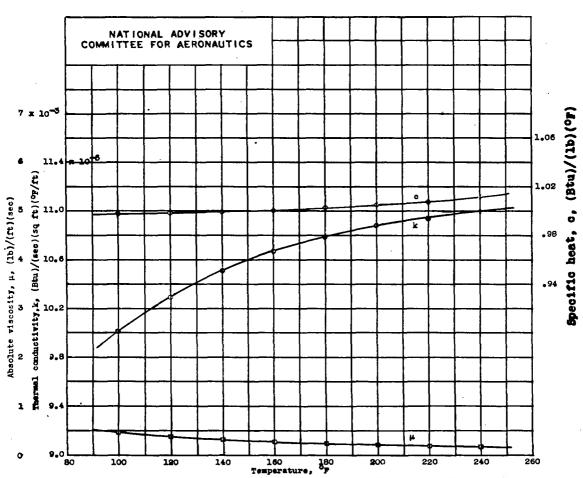


Figure 6. - Variation of specific heat c, thermal conductivity k, and absolute viscosity μ of water with temperature. (Data from reference 2.)

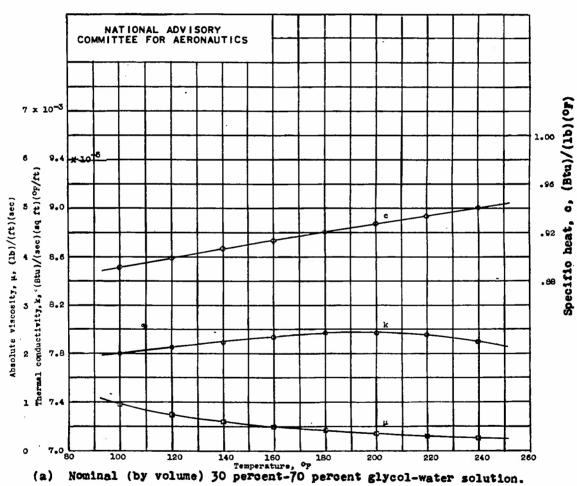
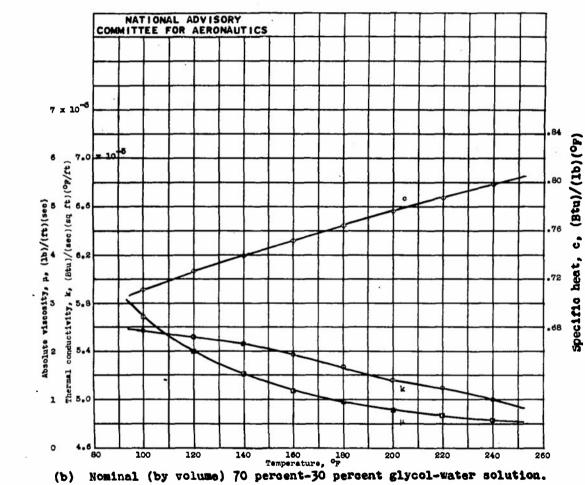
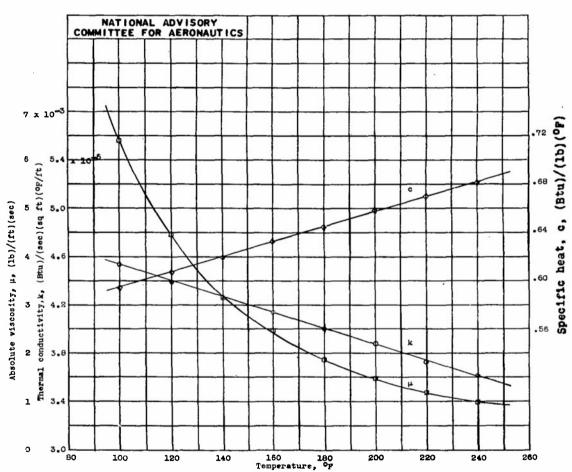


Figure 7. - Variation of specific heat c, thermal conductivity k, and absolute viscosity μ of aqueous ethylene glycol solutions with temperature. (Data from reference 2.)



(b) Nominal (by volume) 70 percent-30 percent glycol-water solution. Figure 7. - Continued.



(c) Nominal (by volume) 97 percent-3 percent glycol-water solution.

Figure 7. - Concluded.

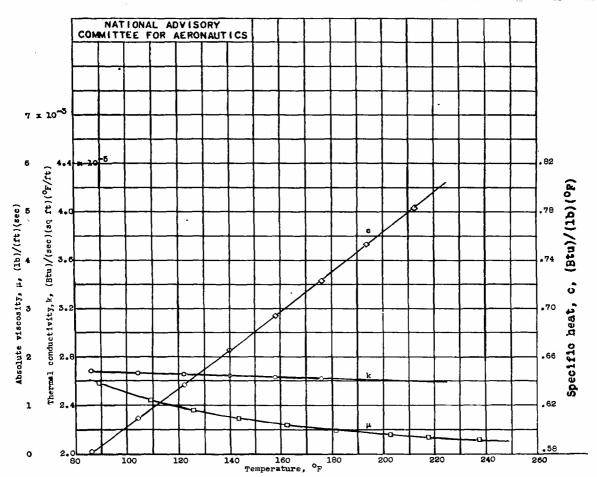


Figure 8. - Variation of specific heat c, thermal conductivity k, and absolute viscosity μ of butanol with temperature. (Data from references 6, 7, and 8.)

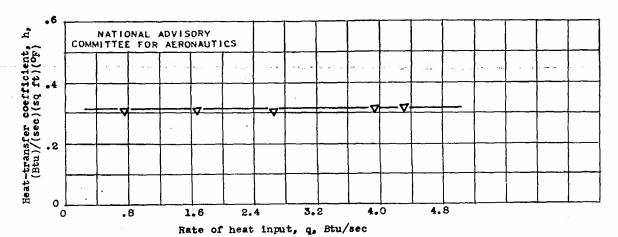


Figure 9.- Variation of heat-transfer coefficient with rate of heat input. Liquid, water; liquid-flow rate, 0.2 pound per second; average liquid temperature, 150° F; liquid pressure, 65 pounds per square inch absolute.

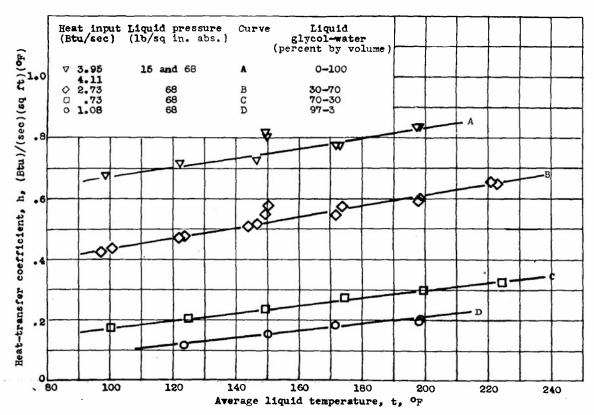
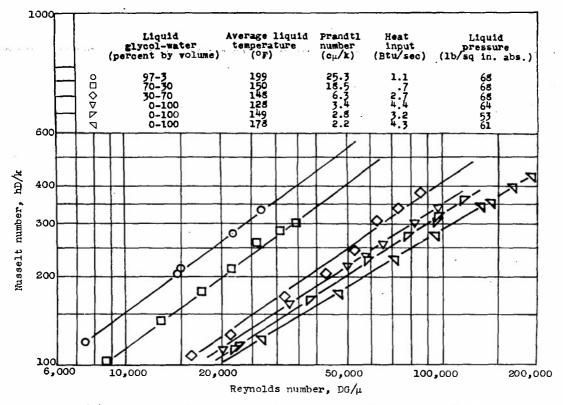
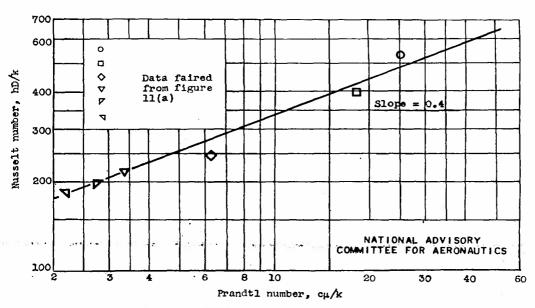


Figure 10. - Variation of heat-transfer coefficient with average liquid temperature for various glycol-water solutions at different approximately constant heat inputs. Liquid-flow rate, 0.67 pound per second.



(a) Variation of Nusselt number with Reynolds number for several liquids at different conditions of operation.



(b) Cross-plot of Musselt number against Prandtl number at a Reynolds number of 50,000 for several liquids at different conditions of operation.

Figure 11.- Determination of exponent n on Prandtl number.

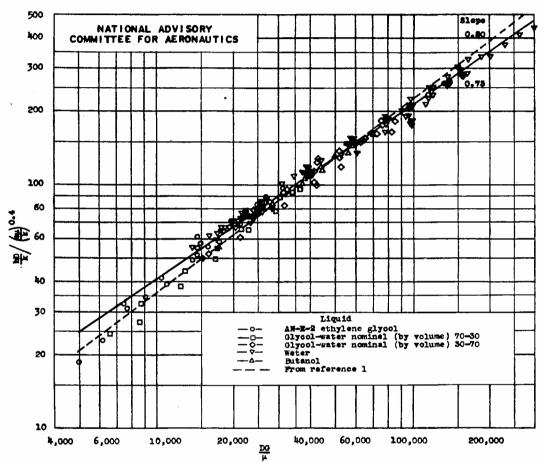
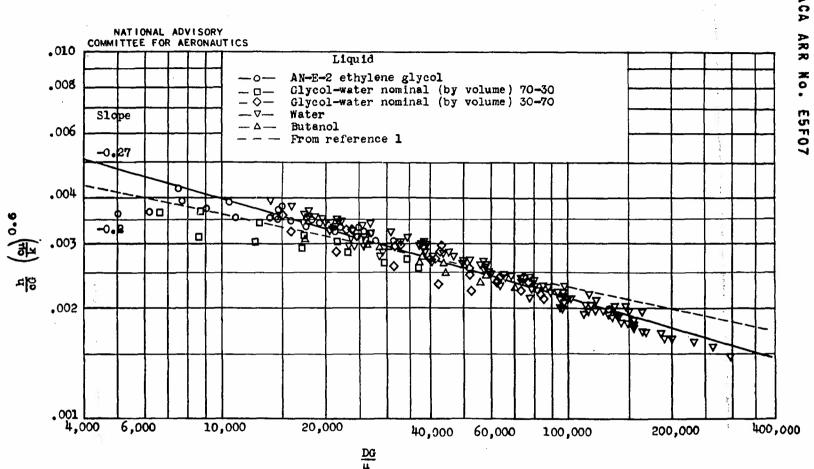


Figure 12.- Correlation of forced-convection heat-transfer data based on Nusselt number for several liquids flowing inside an electrically heated tube under various conditions of average liquid temperature, liquid-flow rate, liquid pressure, end heat input.



 $\frac{\underline{DG}}{\mu}$ Figure 13.- Correlation of forced-convection heat-transfer data based on Stanton number for several liquids flowing inside an electrically heated tube under various conditions of average liquid temperature, liquid-flow rate, liquid pressure, and heat input.

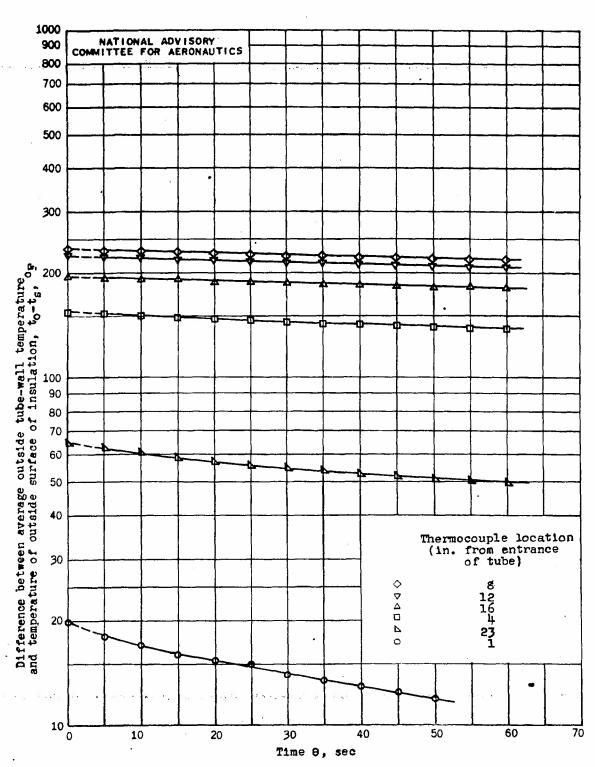
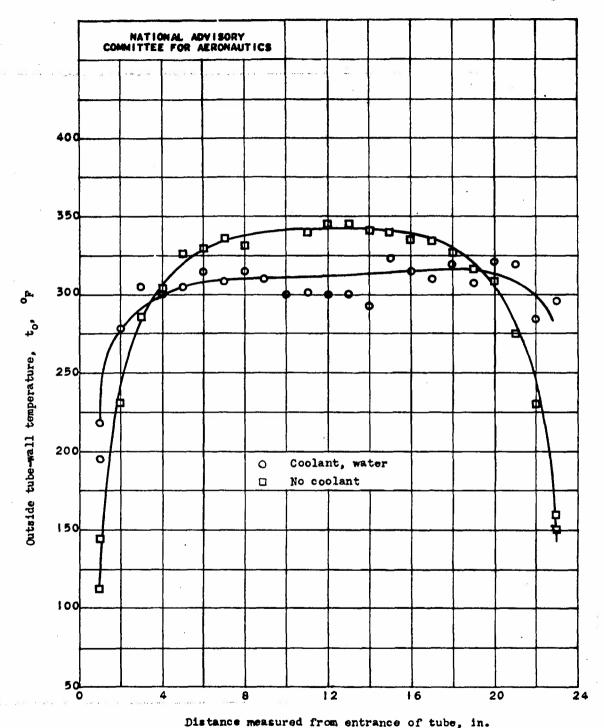


Figure 14.- Cooling-rate curves for several tube-wall thermocouples with no liquid in tube. Temperatures at zero time obtained with power being supplied to tube; temperature of insulation, 95°F.



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Figure 15.- Distribution of outside tube-wall temperatures under two different conditions of operation. With water cooling: average liquid temperature, 122° F; liquid-flow rate, 0.58 pound per second; power input, 1.9 Btu per second. When no coolant was used, power input was set equal to heat loss and all conditions were in equilibrium.

Maca (a) Compac DIVISION: Power Plante, Reciprocating (6) Bernardo, E. DOIG AGENCY NUMBER Eian, Carroll S. SECTION: Cooling (1) ARR-ESFO7 CROSS REFERENCES: Engines - Cooling (32861.99): Coolants - Heat trensfer (27184) DEVISION AUTRODIS) AMED. TITLE Heet-transfer teete of equeous ethylene glycol solutions in en electricelly heeted tube FOOGYN, TITLE-National Advisory Committee for Aeronautice, Weshington, D. C. CONGINATING AGENCY. TRANSLATION. COUNTRY LANGUAGE FORG'NCLASS U. S.CLASS. | DATE PAGES ILLUS. FFATHOUS Unclees, Aug' 45 38 16 table, diagra, graphs v. s. Eng. Tests pere conducted to determine heet transfer characteristics of AN-E-2 ethylene glycol and other ethylene glycol-water mixtures as part of en investigation of the cooling charecteristics of liquid-cooled engines. Liquid temperaturee ranged between 100-250 degrees F. liquid flow rate between 0.17 - 2.50 lb per sec, Reynolde Number between 5000 and 300,000, end heet flux between 4-36 BTU per eec per sq in. In three comparative tests, using varying amounts of glycol-wster, it was found that heat transfer coefficients

of these combinations were 3.8, 2.8 and 1.4 times higher than those of AN-E-2 ethylene glycol.

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